



PALLADIUM CATALYSED REARRANGEMENT OF ALLYLIC SULFOXIMINES: SYNTHESIS OF γ -AMINO α,β -UNSATURATED KETONES AND ESTERS

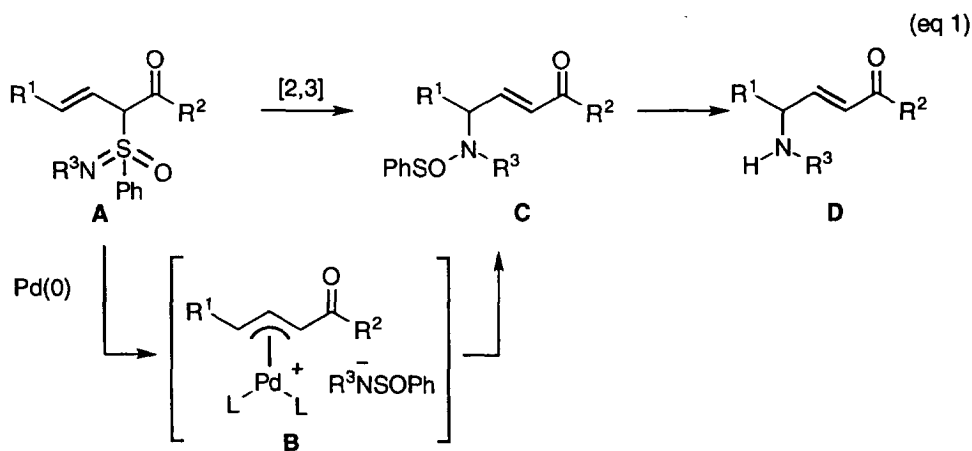
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Abstract: The synthesis of γ -amino α,β -unsaturated ketones and esters from the palladium(0) catalysed rearrangement of (*E*) α -sulfonylimidoyl β,γ -unsaturated ketones and esters is reported.

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γ -Amino α,β -unsaturated ketones and esters are useful substrates for natural product and bioactive molecule synthesis.^{1,2} The latter amino compounds have often been found as important structural elements of peptide-like protease inhibitors.³ γ -Amino α,β -unsaturated esters are readily prepared from the Wittig-Horner reaction of *N*-protected α -amino aldehydes that are available in a few synthetic steps from naturally occurring α -amino acids.^{2,4} This methodology however, is clearly only convenient for the preparation of γ -amino α,β -unsaturated esters from naturally occurring α -amino acids.⁵ As part of a synthetic project we required a general method for the preparation of γ -amino α,β -unsaturated ketones and esters that could not be prepared from "the pool" of naturally occurring α -amino acids. Based on our previous success on the synthesis of chiral allylic amines from the palladium(0) catalysed rearrangement of allylic sulfoximines to allylic sulfinamides⁶⁻⁸ we reasoned that the analogous palladium(0) catalysed rearrangement of α -sulfonylimidoyl β,γ -unsaturated ketones **A** (R^2 = alkyl, aryl) and esters **A** (R^2 = OR) to the allylic sulfinamides **C** would give a route to the desired γ -amino α,β -unsaturated ketones and esters **D** (eq 1). While in principle the thermal [2,3] sigmatropic rearrangement of **A** would give **C** such thermal rearrangements are often inefficient or non-regioselective.⁸⁻¹⁰



The α -sulfonylimidoyl β,γ -unsaturated ketones **3a,b** and ester **3c** were prepared by a Knoevenagel type condensation of the α -sulfonylimidoyl ketones **2a** or **2b** or the known α -sulfonylimidoyl ester **2c**¹¹ with aldehydes as shown in equation 3. The α -sulfonylimidoyl ketones **2a** and **2b** were conveniently prepared according to equation 2 via an aldol like condensation of the carbanions derived from the *S*-methyl sulfoximines **1a**¹² and **1b**¹¹ with benzaldehyde followed by Jones oxidation of the resulting diastereomeric mixture of carbinol compounds. The Knoevenagel type condensation reactions proceeded in modest to good yields (46-87 %) and gave the desired (*E*) α -sulfonylimidoyl β,γ -unsaturated ketones **3a** and **3b** and the (*E*) α -sulfonylimidoyl β,γ -unsaturated ester **3c** as mixtures of two diastereomeric compounds (Table 1).

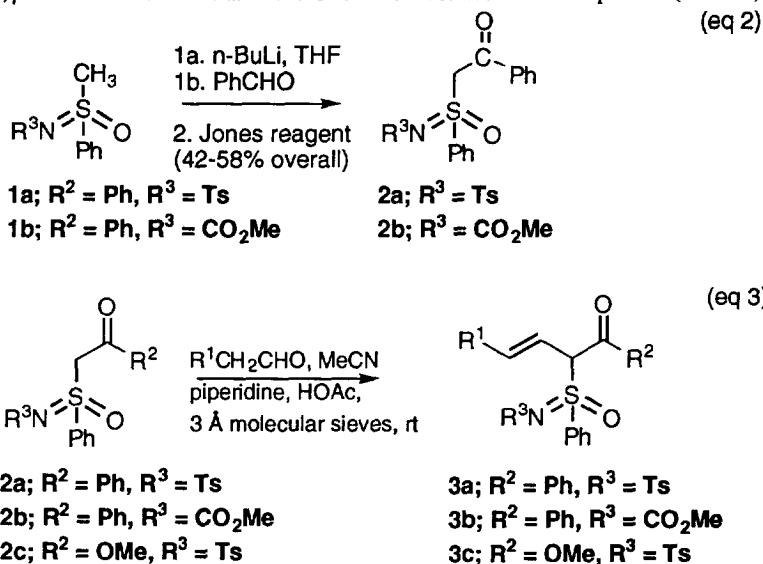


Table 1. Synthesis of **3a-c**

aldehyde (R^1)	product	reaction time (h) ^a	yield(%) ^b	d. r. ^c
n-Bu	3a ; $R^1 = \text{n-Bu}$	4.5	47	74 : 26
n-Bu	3b ; $R^1 = \text{n-Bu}$	24	46	76 : 24
n-pent	3a ; $R^1 = \text{n-pent}$	6	53	76 : 24
n-pent	3b ; $R^1 = \text{n-pent}$	24	65	69 : 31
n-hexyl	3a ; $R^1 = \text{n-hexyl}$	5	53	88 : 12
Et	3c ; $R^1 = \text{Et}$	5	87	58 : 42

^a Not optimised. ^b After purification by column chromatography. ^c Determined by ¹H NMR

Treatment of the individual (*E*) α -sulfonylimidoyl β,γ -unsaturated ketones **3a** or **3b** or the ester **3c** with 10 mol % of freshly prepared tetrakis(triphenylphosphine)palladium(0) ((PPh₃)₄Pd) in dry THF solution at room temperature gave a red or orange coloured solution. TLC analysis of the reaction mixtures after 1h indicated complete consumption of the starting allylic sulfoximines. ¹H NMR analysis of the crude reaction mixtures showed the formation of the often unstable allylic sulfinamides **4a-c**. In the case of the *N*-Ts allylic

sulfenamides **4a** and **4c** ($R^3 = Ts$) these appeared as single diastereomeric products while in the case of **4b** ($R^3 = CO_2Me$) mixtures (75-85 : 25-15) of diastereomeric products were evident. Mild methanolysis of the reaction mixtures with triethylamine / methanol at rt gave pure (*E*)-sulfenamides **5a,c** and the (*E*)-carbamate **5b** after purification of the crude reaction mixtures by column chromatography (silica gel) in overall yields of 32-68 % as shown in Table 2.

We have briefly examined the thermal rearrangement of **3a** and **3b** in acetonitrile at 70-75°C. While the former substrates do undergo rearrangement to **4a** the latter compounds give a complex mixture of products. The extension of this methodology to the asymmetric synthesis of γ -amino α,β -unsaturated ketones and esters and the application of these substrates to the asymmetric synthesis of bioactive molecules is currently under active investigation.

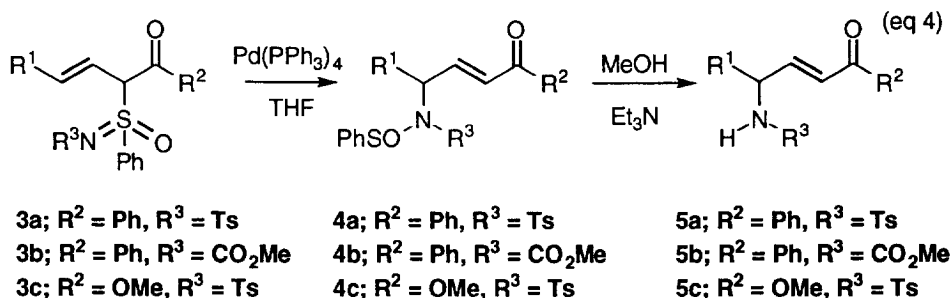


Table 2. Synthesis of **5a-c** from **3a-c**.

starting compound	product	yield(%) ^a	mp (°C)
3a; $R^1 = n\text{-Bu}$	5a; $R^1 = n\text{-Bu}$	32	103-104
3b; $R^1 = n\text{-Bu}$	5b; $R^1 = n\text{-Bu}$	64	oil
3a; $R^1 = n\text{-pent}$	5a; $R^1 = n\text{-pent}$	60	99
3b; $R^1 = n\text{-pent}$	5b; $R^1 = n\text{-pent}$	49	oil
3a; $R^1 = n\text{-hexyl}$	5a; $R^1 = n\text{-hexyl}$	68	ND
3c; $R^1 = Et$	5c; $R^1 = Et$	57	oil

^a After purification by column chromatography.

Experimental

The synthesis of the α -sulfonimidoyl β,γ -unsaturated ketone 3b ($R = n\text{-Bu}$), *a general procedure*: To a stirred mixture of the sulfoximine **1b** (0.269 g, 0.85 mmol), hexanal (0.2 mL, 1.66 mmol) and 3 Å molecular sieves (ca 1 g) in acetonitrile was added a solution of piperidine (18 μ L, 0.18 mmol) and acetic acid (21 μ L, 0.36 mmol) in acetonitrile (3 mL). The mixture was stirred at rt for 24h under an atmosphere of nitrogen. The cloudy yellow solution was then filtered and the solvent was removed *in vacuo*. Purification of the crude product on a short column of silica gel using initially 5% ethyl acetate / hexane and finally 10% ethyl acetate / hexane as eluent gave the title compound as a yellow oil (155 mg, 46 %) and as a 76 : 24 mixture of diastereoisomers. ¹H NMR (CDCl₃, 300 MHz) δ 8.2-7.2 (m, 10H), 7.07 (d, $J = 9.3$ Hz, H1, major diast.),

6.41 (d, $J = 9.3$ Hz, H1, minor diast.), 6.1-5.9 (m, 1H), 5.3-5.4 (m, 1H), 3.80 (s, OMe, major diast.), 3.51 (s, OMe, minor diast.), 2.1-0.7 (m, 9H).

The synthesis of the (E)-carbamate 5b (R = n-Bu), a general procedure. To a solution of the α -sulfonimidoyl β,γ -unsaturated ketone **3b** (R = n-Bu, 0.124 mg, 0.31 mmol) in dry THF (20 mL) was added tetrakis(triphenylphosphine)palladium(0) (0.035 g, 0.031 mmol). The solution was stirred at rt under an atmosphere of nitrogen for 1 h. The solvent was removed *in vacuo* and the yellow orange residue was dissolved in methanol (10 mL). Triethylamine (5 drops) was added and the solution was stirred for 30 min. The solvent was removed *in vacuo*. Purification of the crude product on a short column of silica gel using initially 5% ethyl acetate / hexane and finally 10% ethyl acetate / hexane as eluent gave the title compound as a yellow oil (55 mg, 64 %). ^1H NMR (CDCl_3 , 300 MHz) δ 8.0-7.4 (m, 5H), 7.00 (dd, $J = 15.6, 0.6$ Hz, 1H), 6.89 (dd, $J = 15.6, 5.2$ Hz, 1H), 4.93 (br d, NH , 1H), 4.43 (br s, CHN , 1H), 3.61 (s, CO_2Me , 3H), 2.0-0.8 (m, 9H). ^{13}C NMR (CDCl_3 , 75 MHz) δ 190.5 (CO), 156.4 (CO), 148.1 (CH), 137.6 (C), 132.8 (CH), 128.54 (CH), 128.51 (CH), 124.9 (CH), 52.5 (OMe), 52.3 (CHN), 34.4 (CH_2), 27.7 (CH_2), 22.3 (CH_2), 13.8 (Me).

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